Data collection

Siemans SMART diffractom-	4541 independent reflections
eter	3191 reflections with
Area-detector ω scans	$I > 2\sigma(I)$
Absorption correction:	$R_{\rm int} = 0.053$
multi-scan (Blessing,	$\theta_{\rm max} = 25^{\circ}$
1995)	$h = -7 \rightarrow 7$
$T_{\rm min} = 0.982, T_{\rm max} = 0.995$	$k = 0 \rightarrow 12$
14889 measured reflections	$l = 0 \rightarrow 44$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0566P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.057$	+ 0.6084 <i>P</i>]
$wR(F^2) = 0.135$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.076	$(\Delta/\sigma)_{\rm max} = 0.013$
4541 reflections	$\Delta \rho_{\rm max} = 0.184 \ {\rm e} \ {\rm \AA}^{-3}$
317 parameters	$\Delta \rho_{\rm min} = -0.210 \ {\rm e} \ {\rm \AA}^{-3}$
H atoms treated by a	Extinction correction: none
mixture of independent	Scattering factors from
and constrained refinement	International Tables for
and constrained remember	Crystallography (Vol. C)

Table 2. Selected geometric parameters (Å, °) for (II)

C4—C5 C5—C10 C8—C9 C8—C14	1.562 (5) 1.559 (4) 1.553 (4) 1.568 (4)	C9—C10 C13—C18 C13—C14	1.570 (4) 1.548 (4) 1.575 (4)
C6-C5-C10	111.1 (3)	C8—C9—C10	117.3 (2)
C6-C5-C4	115.2(3)	C15 - C14 - C8	113.4 (2)
C10-C5-C4	116.9(3)	C15—C14—C13	111.7 (3)
C11-C9-C8	110.0(2)	C8-C14-C13	116.7 (2)
C11-C9-C10	114.3 (3)		

Table 3. Hydrogen-bonding geometry (Å, °) for (II)

$D - H \cdot \cdot \cdot A$	D—H	$\mathbf{H} \cdot \cdot \cdot \mathbf{A}$	$D \cdot \cdot \cdot A$	$D = H \cdots A$
04—H4A···O5	0.83	1.97	2.666 (4)	140.6
O5—H5· · ·O3 ⁱ	1.20 (5)	1.59 (5)	2.763 (4)	167 (3)
Symmetry code: (i	1 - x, y - 1	$\frac{1}{3} - z$.		

The data collection nominally covered over a hemisphere of reciprocal space by a combination of three sets of exposures; each set had a different φ angle for the crystal and each exposure covered 0.3° in ω . The crystal-to-detector distance was 4.94 cm. In each case a total of 1271 frames was collected. Crystal decay was monitored by repeating the initial frames at the end of data collection and analysing the duplicate reflections. Crystals of (I) were of poor quality, yielding a weak diffraction pattern and leading to a higher final residual.

H atoms were placed geometrically and refined with a riding model (including free rotation about C-C bonds for methyl groups), and with U_{iso} constrained to be 1.2 (1.5 for methyl groups) times U_{eq} of the carrier atom. The scalarin structure (II) also contains a methanol of solvation in which the hydroxyl-H atom was located from a difference map.

The absolute configuration could not be determined in either case and the figures depict the same absolute configuration as scalarane (Kazlauskas et al., 1980).

For both compounds, data collection: SMART (Siemens, 1994); cell refinement: SAINT (Siemens, 1994); data reduction: SAINT; program(s) used to solve structures: SHELXS97 (Sheldrick, 1990); program(s) used to refine structures: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Siemens, 1994); software used to prepare material for publication: SHELXL97.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SX1072). Services for accessing these data are described at the back of the journal.

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9-Acetonyl-3-ethylidene-1,2,3,4-tetrahydrospiro[carbazole-1,2'-[1,3]dithiolan]-4-one

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Abstract

The title compound, $C_{18}H_{19}NO_2S_2$, consists of a carbazole skeleton with a pentacyclic dithiolane ring spirobonded at position 1 and acetonyl and ethylidene chains at positions 9 and 3, respectively. The heteroatoms and the acetonyl and ethylidene chains in the molecule are responsible for the changes in the bond lengths and angles of the carbazole core.

Comment

The title compound can be considered as a synthetic precursor of a tetracyclic indole alkaloid, dehydro-de-(N-methyl)uleine, which has been isolated from Aspidosperma (Joule et al., 1965). The integration of the amino-functionalized group on the C4 atom (according to the carbazole numbering) following the ring cyclization could constitute a new synthetic entry in to the total synthesis of dehydro-de-N-methyluleine, similar to the synthesis of the tetracyclic indole alkaloids in the dasycarpidone series (Patır, 1995; Patır et al., 1996). The structure determination of the title compound, (I), was undertaken in order to understand the effects of the acetonyl and ethylidene groups at positions 9 and 3, respectively, on the geometry of the carbazole system and to compare the obtained results with those of spiro[carbazole-1(2H),2'-[1,3]dithiolan]-4-one (Hökelek et al., 1998).



The title compound consists of a carbazole skeleton, a pentacyclic dithiolane ring spiro-bonded at position 1 and acetonyl and ethylidene chains at positions 9 and 3, respectively. The interatomic distances N9 \cdots H8 2.77 (2), S1 \cdots H122 2.69 (2), S1 \cdots H21 2.69 (2), S2 \cdots H22 2.83 (2), O1 \cdots H5 2.66 (2), O1 \cdots H141 2.482 (4) and O2 \cdots H102 2.56 (2) Å indicate the close contacts causing changes in the values of bond lengths and angles of the carbazole skeleton.

The C—N bonds [N9—C8a 1.396(2) and N9— C9a 1.377(2)Å] are nearly the same as the corresponding values [N9—C8a 1.382(2) and N9—C9a 1.356(3)Å] in spiro[carbazole-1(2*H*),2'-[1,3]dithiolan]-4-one (Hökelek *et al.*, 1998), but shorter than the corresponding values [N9—C8a 1.423(5) and N9—C9a 1.412(5)Å] in 2,3-dihydro-9-(phenylsulfonyl)carbazole-4(1*H*)-one (Hökelek *et al.*, 1994), even though there is a -CH₂-O—CH₃ protecting group at the N9 position. The -CH₂-O—CH₃ protecting group and -CH₂-CH₃ chain at the N9 and C3 positions cause notable conformation changes on the geometry of the carbazole core leading to angle size increases [C1-C2-C3 114.3 (2), C4-C4a-C9a 124.5 (2), C9a-C1-S1 $115.9(1), O1-C4-C3 122.9(2)^{\circ}$ and decreases [C3-C4-C4a 114.6(2), C4-C4a-C5a 127.5(2), C4a-C9a-C1 122.8 (2), C9a-N9-C8a 108.1 (1), C9a-C1—S2 106.7 (1), C2—C1—S1 106.3 (1) $^{\circ}$] according to the corresponding values [112.3(2), 122.0(2), 112.8(1), 121.5(2), 115.9(2), 130.9(2), 124.3(2),109.6(1), 108.0(1) and $109.5(1)^{\circ}$, respectively] in spiro[carbazole-1(2H),2'-[1,3]dithiolan]-4-one (Hökelek et al., 1998). An examination of the deviations from the least-squares planes through the individual rings shows that rings A (C5a, C5, C6, C7, C8, C8a) and B (C4a, C5a, C8a, N9, C9a) are planar. The C (C1, C2, C3, C4, C4a, C9a) and D (C1, S1, C10, C11, S2) rings are not planar with maximum deviations at C2 [0.315(2)] and C11 $\left[-0.487(2) \text{ Å}\right]$. They are also twisted with respect to each other. The dihedral angles between the best leastsquares planes are $A^{\wedge}C = 4.91(4), A^{\wedge}D = 83.78(5),$ $B^{\wedge}C = 4.51$ (6), $B^{\wedge}D = 82.79$ (5) and $C^{\wedge}D = 80.77$ (5)°. Ring C has a sofa conformation with a local pseudo mirror running along C2...C4a, while the conformation of ring D is half-chair with a local pseudo twofold axis running along S1 and the midpoint of the S2-C11 bond.



Fig. 1. An ORTEPII (Johnson, 1976) drawing of (1) with the atomnumbering scheme. Displacement ellipsoids are drawn at the 50% probability level for non-H atoms.

Experimental

The title compound, (I), was prepared from the reaction of lithium bis(trimethylsilyl)amide (15 mmol) in 30 ml tetrahydrofuran and 9-methoxymethyl-2,3-dihydrospiro[1*H*carbazole-1, 2'-[1,3]dithiolane]-4-one (4.0 g, 12.52 mmol) (Masamune *et al.*, 1982). The resulting mixture was stirred at 223 K for 2 h and then acetaldehyde (1 ml, 18 mmol) was added and stirring continued overnight. The mixture was then poured into HCl solution (50 ml, 10%), extracted with chloroform, dried over MgSO₄ and evaporated. The title compound %

(I), was obtained from column chromatography [m.p. 404 K, 3.2 g (74%) yield] and recrystallized from ethyl acetate.

Crystal data

 $C_{18}H_{19}NO_2S_2$ Mo $K\alpha$ radiation $M_r = 345.49$ $\lambda = 0.71073 \text{ Å}$ Orthorhombic Cell parameters from 25 reflections Pbca a = 9.202(1) Å $\theta = 9 - 11^{\circ}$ $\mu = 0.324 \text{ mm}^{-1}$ b = 18.870(3) Å c = 19.138(2) Å T = 298 K $V = 3323.2(7) \text{ Å}^3$ Block $0.30 \times 0.25 \times 0.20$ mm Z = 8 $D_x = 1.381 \text{ Mg m}^{-3}$ Yellow D_m not measured

Data collection

Enraf-Nonius CAD-4
diffractometer
$$R_{int} = 0.023$$

 $\theta_{max} = 26.3^{\circ}$
 $h = 0 \rightarrow 11$ $\omega/2\theta$ scans $h = 0 \rightarrow 11$ Absorption correction:
empirical via ψ scans
 $(MolEN;$ Fair, 1990) $l = 0 \rightarrow 23$
 3 standard reflections
 $rmin = 0.910, T_{max} = 0.940$ 3816 measured reflections
 3363 independent reflections
 2922 reflections with
 $F > 2\sigma(F)$ $R_{int} = 0.023$
 $\theta_{max} = 26.3^{\circ}$
 $h = 0 \rightarrow 11$
 3 standard reflections
every 230 reflections
intensity decay: 1%

Refinement

Refinement on F	$(\Delta/\sigma)_{\rm max} = 0.01$
R = 0.038	$\Delta \rho_{\rm max} = 0.34 \ {\rm e} \ {\rm \AA}^{-3}$
wR = 0.048	$\Delta \rho_{\rm min} = -0.21 \ {\rm e} \ {\rm \AA}^{-3}$
S = 1.22	Extinction correction: none
2922 reflections	Scattering factors from
256 parameters	International Tables for
H atoms: see text	X-ray Crystallography
$w = 1/[\sigma(F)^2 + (0.02F)^2]$	(Vol. IV)
+ 1.0]	

Table 1. Selected geometric parameters (Å, °)

		-	
\$1—C1	1.856 (2)	C3C4	1.495 (3)
\$1—C10	1.815(2)	C4—C4a	1.449 (3)
\$2—C1	1.832 (2)	C4a—C5a	1.439 (3)
\$2—C11	1.799 (2)	C4a—C9a	1.384 (2)
01—C4	1.224 (2)		
C1S1C10	97.89 (9)	C3-C4-C4a	114.6 (2)
C1-S2-C11	95.04 (9)	C1—C9a—C4a	122.8 (2)
\$1—C1—S2	107.37 (9)	S1-C10-C11	109.0 (2)
\$2—C1—C2	112.1(1)	S2-C11-C10	107.2 (1)
C2-C1-C9a	108.6(1)	C4—C4a—C9a	124.5 (2)
C1-C2-C3	114.3 (2)	C8aN9C9a	108.1(1)
C2—C3—C4	115.1(2)		

Most of the H-atom positions were located by difference synthesis and refined isotropically. The remaining ones were positioned geometrically and a riding model was used during the refinement process.

Data collection: MolEN (Fair, 1990). Cell refinement: MolEN. Data reduction: MolEN. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: MolEN. Molecular graphics: ORTEPII (Johnson, 1976). Software used to prepare material for publication: MolEN.

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4-(2,5-Di-tert-butylphenylnitrilomethylidyne)-2-methoxyphenol

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Abstract

The title compound, $C_{22}H_{29}NO_2$, is a monodentate Schiff base. There are two independent molecules in the asymmetric unit. In each molecule the two planar phenyl rings bridged by a C==N imino moiety are nearly perpendicular to one another $[79.80(4) \text{ and } 81.14(4)^{\circ}]$. Each molecule contains intramolecular $(O - H \cdot \cdot \cdot O)$ and intermolecular (O— $H \cdots N$) hydrogen bonds.

Comment

Schiff bases and their transition metal complexes have been extensively studied during the last decade.